Electronic, vibrational, and thermodynamic properties of metacinnabar β -HgS, HgSe, and HgTe

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We report *ab initio* calculations of the electronic band structure and the phonon dispersion relations of the zincblende-type mercury chalcogenides (β -HgS, HgSe, and HgTe). The latter have been used to evaluate the temperature dependence of the specific heat, which has been compared with experimental data. The electronic band structure of these materials has been confirmed to have an inverted direct gap of the α -tin type, which makes HgSe and HgTe semimetallic. For β -HgS, however, our calculations predict a negative spin-orbit splitting that restores semiconducting properties to the material in spite of the inverted gap. We have calculated the spin-orbit induced linear terms in **k**, which appear at the Γ_8 valence bands. We have also investigated the pressure dependence of the crystal structure and the phonons.

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I. INTRODUCTION

The present work is part of an ongoing effort to investigate, both theoretically and experimentally, the temperature dependence of the specific heat $C_v \approx C_p$ of semiconductors and semimetals in the low temperature region in which strong deviations from the Debye T^3 law take place.¹ These deviations lead to a peak in C_p/T^3 vs T at a temperature T $\approx \Theta/20$, where Θ is the Debye temperature. Previous investigations have focused on diamondlike semiconductors and their zincblende and wurtzite analogs.²⁻⁶ More recently, the lead chalcogenides, with rock-salt structure, have been investigated.⁷ They, as well as the structurally related semimetals Bi and Sb (cf. Ref. 8), exhibit a remarkable dependence of their vibrational properties, and, correspondingly, C_v , on spin-orbit (s-o) interaction.^{9,10} C_v calculations, and the corresponding C_p measurements, have also been reported for the related semiconductor (rhombohedral structure) GeTe, a material that grows with a large hole concentration that enables the investigation of the effect of such doping on the vibrational properties and on C_p (or C_v).¹¹ This work usually included measurements on samples prepared with different isotopic compositions, so as to investigate their effect on C_p , accompanied by similar calculations of C_v .

Because of the calculated dependence of the aforementioned properties on s-o interaction for materials composed of heavy atoms (e.g., Bi, Pb) and structures related to rock salt, we decided to investigate similar materials that crystallize in the zincblende structure. The mercury chalcogenides appeared to be ideal for this purpose, although the toxic nature of mercury dissuaded us from investigating isotope effects. Moreover, the seven stable isotopes of Hg have low isotopic abundances and correspondingly prohibitive prices. These materials, however, have a number of hitherto not fully investigated electronic properties, related to the s-o interaction and to the gap inversion, which makes of them zero-gap semiconductors.¹² We have investigated some of these properties in the process of calculating *ab initio* the phonon dispersion relations and C_v .

HgS is usually found in nature as cinnabar, crystallized in the Se-like trigonal structure (α -HgS), a structure in which it usually also appears when grown in the laboratory. A small addition of Fe helps to crystallize HgS in the zincblende structure $(\beta$ -HgS), both in nature [(~1%)Fe]. metacinnabar,¹³ and in the laboratory.¹⁴ We use for the present work natural crystals of β -HgS originating from Mount Diablo Mine, Clayton, CA (purchased from John Betts, New York, NY 10025). Black A chemical microanalysis of two such crystals taken from different parts of the sample gave an Fe content of ~ 1.1 mass-% Fe. Additionally, we measured the specific heat of a commercially available polycrystalline fine powder of 99.999% stated purity (Alfa Aesar GmbH & Co KG, Karlsruhe). HgSe and HgTe can also be found in nature (HgSe: Tiemannite, HgTe: Coloradoite) but they are rare. We used for the work reported here synthetic crystals grown by the Bridgman or sublimation technique, respectively. Cinnabar is transparent in the red and has an optical gap of ~ 2.275 eV.¹⁵ We shall not discuss it here any further since we confine ourselves to the chalcogenides with zincblende structure. It will be discussed in a later publication. The zincblende variety of HgS (β -HgS, metacinnabar), which will be fully discussed here, has a black color and a zero or near zero gap. All crystals used here were found to be *n*-type by thermoelectric power measurements.

The structure of this article is as follows: in Sec. II, we discuss the procedures employed for the *ab initio* calculations of the electronic band structure, the phonon dispersion relations and the heat capacity. Section III presents some important results obtained in the process of performing the band structure calculations. We display the full LDA electronic band structure of β -HgS, which has not been reported

TABLE I. Values at $\mathbf{k}=0$ of the spin-orbit splitting Δ_0 , the negative *p*-s gap E_0 (in eV) and the lattice constant a_0 (in Å) as calculated with various codes and experimentally determined. Notice the negative value of Δ_0 calculated by our three methods for HgS. It is due to an admixture of *p* wave functions of the cation with 5*d* of Hg (the latter have a negative contribution similar to results found for CuCl, ZnO, and CuGaS₂). The differences of Δ_0 for the different computational methods are due to minor differences in the compensating *p* and *d* contributions. The negative Δ_0 of β -HgS has already been mentioned recently (Refs. 22 and 23).

		HgS	HgSe	HgTe
$\Delta_0 (eV)$	VASP-LDA	-0.111	+0.23	+0.783
	ABINIT-LDA	-0.18	+0.34	+0.81
	VASP-GGA	-0.091	+0.25	+0.751
	Overhof		+0.48	+1.13
	Experimental	-0.25 ^a	+0.38 b	0.9 ^b
E_0 (eV)	VASP-LDA	-0.573	-1.18	-1.025
	ABINIT-LDA	-0.58	-1.26	-1.15
	VASP-GGA	-0.483	-1.07	-1.113
	Overhof		-0.4	-0.3
	Experimental	-0.15 ^a	\sim -0.4 ^b	-0.3 ^b
$a_0(\text{\AA})$	VASP-LDA	5.825	6.065	6.433
	ABINIT-LDA	5.80	6.054	6.442
	VASP-GGA	5.999	6.255	6.633
	ABINIT-GGA	5.982	6.250	6.656
	Overhof		6.075	6.45
	Experimental	5.841(~295 K) ^c	6.085 (293 K) ^b	6.453 ^b

^aReference 13.

^bReference 25.

^cReference 26.

in the literature so far. We have found, however, three articles discussing the band structures of HgSe and HgTe, which make it unnecessary to reproduce here our full calculated band structures for these materials. The oldest, by Bloom and Bergstresser,¹⁶ uses an empirical pseudopotential method that includes s-o interaction. The second one, by Overhof,¹⁷ uses the Korringa-Kohn-Rostocker (KKR) method. In spite of not being self-consistent, these calculations already reveal the inverted, α -tin-like band structure of these materials, as it does the self-consistent linear muffin-tin orbital (LMTO) calculation for HgTe performed by Cade and Lee.¹⁸ In the process of analyzing our band structure calculations, we realized that there is a considerable contribution of the core 5d levels of Hg to the s-o splitting at the Γ point of the valence bands. Like in the case of the copper monohalides, this contribution has a sign opposite to that of the splitting of the *p* valence electrons of the anion.^{19,20} For HgS, this 5d contribution overcompensates that of the 3p states of sulfur, thus reversing the corresponding valence states and changing the topology of the bands. This situation is similar to that found in CuCl, ZnO, and CuGaS₂.²¹ The calculated s-o splittings for the three Hg compounds are given in Table I. This table also contains calculated values of the negative Γ_8 - Γ_1 gap and the lattice parameter a_0 obtained through energy minimization. Experimental data for these parameters are also given.

Another interesting aspect of the band structures of the HgS, HgSe, and HgTe is the appearance of linear terms in the wave vector \mathbf{k} at the Γ_8 point of the valence bands. Such terms, of relativistic origin, were predicted by Dresselhaus for zincblende-type materials on symmetry grounds.²⁷ They arise from *s*-*o* coupling plus the lack of inversion symmetry and involve $\mathbf{k} \cdot \mathbf{p}$ interactions between the valence bands and the core *d* electrons.²⁸ Very few *ab initio* calculations of these terms, which are important in present day spintronics, have been performed and the Hg-chalcogenides should be ideal materials for such calculations. We thus present enlarged versions of the valence band structures around Γ and use them to evaluate the coefficients of these linear terms. The results are presented in Table II.

Section IV is devoted to the *ab initio* lattice dynamical calculations and comparison with inelastic neutron scattering and Raman data. For the purpose of interpreting the specific heat results, we also present the corresponding one-phonon densities of states and their decomposition in Hg and anion contributions. The possible contributions to the linewidths of the Raman phonons, anharmonicity, electron-phonon interaction, and isotopic disorder is also discussed. For completeness, we also present the density of two-phonon states (sum and difference) corresponding to a total zero \mathbf{k} vector. These results are useful for the interpretation of two-phonon Raman

TABLE II. Values of the coefficients of the linear terms around the Γ_8 states (in meV Å) of the valence band structures of β -HgS, HgSe, and HgTe as calculated with the VASP and ABINIT codes through a quadratic fit of the eigenvalues (see Fig. 3).

	<i>β</i> -ŀ	β-HgS		HgSe		HgTe	
	$C_k[100]$	$C_k[111]$	$C_k[100]$	$C_k[111]$	$C_k[100]$	$C_k[111]$	
VASP-GGA	-147	-140	-145	-146	-145	-137	
ABINIT-LDA	-144	-174	-160	-170	-148	-136	

scattering data; they are compared with spectra available in the literature for HgSe.

Because of the lack of theoretical information concerning the phase transitions of the Hg-chalcogenides under pressure, we have also performed calculations of the enthalpy of HgS and HgTe vs pressure for the zincblende, rock-salt, and cinnabar phases. From the calculated dependence on pressure p, the values of p at which transitions occur are obtained and compared with experimental results. As a by-product, we report calculated values of the Grüneisen parameters of the LO and TO phonons of HgTe at the Γ point of the Brillouin zone. This work is presented in Sec. V.

Section VI discusses the specific heats and their temperature dependence around the maximum of C_v/T^3 . The *ab ini*tio calculations, which have been performed with and without s-o interaction and using either the LDA or the GGA approximation, describe rather well the experimental results. The inclusion of s-o interaction lowers the maximum of C_n/T^3 by about 8% in the three compounds with zincblende structure. Since this effect has a sign opposite to that found for the rock-salt structure semiconductors (e.g., PbS), we tried to ascertain whether the sign of the effect depends on crystal structure. For this purpose, we calculated the lattice dynamics and C_v of HgS in the pressure region where the rock-salt phase is the stable one. We found a very small effect of the s-o interaction at the maximum of C_{ν}/T^3 , corresponding to a decrease by less than 1%. Section VII summarizes the conclusions.

II. THEORETICAL DETAILS

In this paper, we study the vibrational and thermal properties of HgX (X=S, Se, and Te) by three different approaches: an experimental one and two different calculations based on different implementation of density functional theory.²⁹ For the theoretical part, we have used two different computer codes: ABINIT and VASP.^{30,31} In order to describe the exchange and correlation effects, we have used two wellknown approximations, local density approximation^{32,33} (LDA) and Perdew, Burke, Ernzerhof generalized gradient approximation (PBE-GGA).³⁴

In the ABINIT calculations, the wave function is expanded in plane waves and a norm-conserving pseudopotential approximation to the free ion potential is employed, with the Teter-Padé parametrization for the exchange and correlation.^{35,36} In particular, we did use the Hartwigsen-Goedecker-Hutter (HGH) pseudopotentials, where scalarrelativistic and spin-orbit (*s-o*) terms are combined, such that they fully represent nonlocal pseudopotentials of separable form.^{35,37,38} A 60 Ha kinetic energy cutoff was applied and a Monkhorst-Pack grid of $6 \times 6 \times 6$ was used in order to describe any electronic quantity within the Brillouin zone. Electronic and vibrational properties were checked with respect to these parameters to ensure convergence. The lattice dynamics was calculated by using an implementation of density functional perturbation theory.³⁹ Vibrational frequencies and the vibrational density of states were calculated after the diagonalization of the interatomic force constants (IFC). These were used to obtain the thermal properties within the harmonic approximation. An original grid of $12 \times 12 \times 12$ within the Brillouin zone, with a total of 72q points, was used to calculate the corresponding set of IFCs. Afterwards, a finer grid was employed to determine the thermal properties, obtained by performing a Fourier fit to the calculated IFCs (for example, in order to calculate C_v , a set of 32768q points was used, which was reduced by symmetries to 897 points).

The Vienna ab initio simulation package (VASP) is a density functional theory implementation with pseudopotentials used to obtain the electron wave functions (see Ref. 31 and references therein). The exchange-correlation energy was taken in the local density approximation with the Ceperley Alder prescription. Calculations with the generalized gradient approximation (PBE-GGA) were also performed.33,34 The projector-augmented wave (PAW) scheme was adopted and the semicore 5d and 6s electrons of Hg were included explicitly in the calculations.²⁴ The set of plane waves used extended up to a kinetic energy cutoff of 370 eV for HgS and 310 eV for HgTe and HgSe. This large cutoff was required in order to achieve highly converged results within the PAW scheme. We used a dense Monkhorst-Pack grid for Brillouin Zone (BZ) integrations in order to ensure highly converged results in the three compounds (to about 1 and 2 meV per formula unit). We also used an accurate algorithm during the calculations which allowed us to obtain highly converged forces for the calculation of the dynamical matrix. At each selected volume, the crystal structure was fully relaxed to its equilibrium configuration through the calculation of the forces on atoms and the stress tensor.⁴⁰ In the relaxed equilibrium configuration, the forces are less than 0.002 eV/Å and the deviation of the stress tensor from a diagonal hydrostatic form is less than 0.1 GPa.

Highly converged results on forces are required for the calculation of the dynamical matrix using the direct force constant approach (or supercell method).⁴¹ The construction of the dynamical matrix at the Γ point of the BZ is particu-



FIG. 1. Electronic band structure of β -HgS calculated with the ABINIT code in the LDA approximation (see Sec.II) and with *s*-*o* interaction using the lattice parameter a_0 =5.80 Å obtained by total energy minimization.

larly simple and involves separate calculations of the forces in which a fixed displacement from the equilibrium configuration of the atoms within the primitive unit cell is considered. Symmetry helps to decrease the number of such independent distortions, thus reducing the computational effort in the study of the analyzed structures considered in this work. Diagonalization of the dynamical matrix provides both the frequencies of the normal modes and their polarization vectors, which allow us to identify the irreducible representation and the atomic character of the phonons modes at the relevant k point. Supercell calculations were carried out in order to obtain phonon dispersion curves. Starting with a 2 $\times 2 \times 2$ supercell, the direct method delivers correct harmonic phonon frequencies at the special wave vectors of the Brillouin zone that are compatible within the selected supercell size. If the interaction range would cease at the selected supercell size, the corresponding phonon frequencies would provide a good basis for the interpolation to all wave vectors. The LO/TO splitting cannot be automatically included when using the direct method, but the dynamical matrix can be supplemented with a nonanalytical term, which depends on the Born effective charge tensors and the electronic dielectric constant.⁴² The thermodynamic properties of the crystals are, to a large extent, determined by phonon dispersion and can be satisfactorily described in most of the cases within the quasiharmonic approximation, as discussed in Ref. 41.

III. ELECTRONIC BAND STRUCTURE

Figure 1 displays the band structure of β -HgS as calculated with the ABINIT code using the LDA approximation (see Sec. II) with inclusion of *s*-*o* interaction. The band structure shown in this figure was calculated with the energy optimized lattice parameter a_0 =5.80 Å. The *s*-*o* splitting of the Γ_{15} valence bands (-0.18 eV) cannot be resolved in this figure because of the wide energy scale it covers. It will be discussed later in connection with Fig. 2. The wide energy scale, however, allows us to observe the position of the low-



FIG. 2. Electronic band structure of β -HgS calculated with the ABINIT code in the LDA approximation (see Sec. II) using the lattice parameter a_0 =5.80 Å obtained by total energy minimization as in Fig. 1, but with the energy scale enlarged so as to display the details of the top valence bands, including *s*-*o* splitting, and the lowest conduction band around Γ . The Γ_8 - Γ_7 separation corresponds to the negative *s*-*o* splitting Δ_0 . Γ_6 - Γ_8 is the E_0 gap. The inset represents the band structure without *s*-*o* splitting near the Γ point.

est, sulfur 3*s*-like Γ_1 valence band, which is not affected by *s*-*o* interaction, at about -12 eV. Above this band, the 5*d* bands of Hg appear. Without *s*-*o* interaction they split into an upper quadruplet (~-6 eV, Γ_{12}) and a lower sextuplet (Γ_{15}). *s*-*o* interaction does not split the Γ_{12} states but it splits the Γ_{15} sextuplet into a doublet Γ_7 and a quadruplet Γ_8 , the former being above the latter. The splitting shown in Fig. 1 is 2.1 eV, the same as the corresponding atomic splitting of Hg.⁴³

At -0.58 eV, we find an s-like Γ_1 state, which is composed mainly of 6s states of Hg. This state corresponds to the lowest conduction band for most tetrahedral semiconductors (exception: $\alpha \tan^{12}$ and, as discussed below, HgSe and HgTe) above the Γ_{15} valence bands. In order to resolve the details of the band structure around the Γ point, we replot in Fig. 2 the data from Fig. 1 with an expanded energy scale. This figure shows the inverted s-o interaction at Γ : the Γ_7 doublet is above the Γ_8 quadruplet. This fact arises from the contribution of the negative s-o splitting of the $5d\Gamma_{15}$ states of Hg (Fig. 1) which overcompensates the positive splitting of the S 3p contribution. This effect is similar to that observed for CuCl (Refs. 19 and 20) but for Hg an additional complication arises, related to the α -tin-like structure. Counting bands, we see that the Γ_8 bands should be occupied whereas the *s*-*o* split Γ_7 should be empty. Hence, according to Figure 2, undoped (intrinsic) β -HgS should have an indirect energy gap of ~ 0.15 eV and a direct one of 0.18 eV. This is in reasonable agreement with the gap observed by Zallen and Slade (0.25 eV), although these authors did not realize that the *s*-*o* splitting of β -HgS is negative.¹³

In Table I, we have listed the negative gaps E_0 between the valence bands and the Γ_6 Hg5s-like conduction band calculated and observed for the three Hg-chalcogenides. We notice in the table that the calculated positions of the Γ_6 bands are lower than the experimental ones by 0.43 eV for HgS and



FIG. 3. (Color online) Γ_8 valence bands of HgSe showing the linear terms in **k** along the [100] and the [111] directions, as calculated with the ABINIT-LDA code. The colored curves represent fits to a quadratic polynomial, the black ones connect calculated points as a guide to the eye. The values of the fit coefficients are given in the inset. The zero of energy lies at the valence band maximum at the Fermi energy.

0.85 eV for HgSe and HgTe. It is natural to attribute this fact to the so-called "gap problem" inherent to the LDA calculations.⁴⁴ Recent *GW* show that the Γ_1 state of the HgX materials should be raised by about 0.75 eV.^{45,46} If we add this energy to the values of E_0 displayed in Table I, the E_0 gaps of HgSe and HgTe remain negative (~-0.35 eV), whereas that of β -HgS becomes positive (~+0.4 eV). Thus, β -HgS would be a conventional semiconductor except for the negative Δ_0 . As already mentioned, we do not reproduce here the full band structures of HgSe and HgTe because they have already been presented in the literature.^{16–18} We have given a few important parameters in Table I, also for HgSe and HgTe. An interesting fact is their positive *s-o* splitting Δ_0



FIG. 4. (Color online) Phonon dispersion relations of β -HgS calculated with the ABINIT-LDA code with (solid line) and without (dashed line) *s*-*o* coupling. The circles were obtained with INS (Ref. 50). The LO and TO phonons at $\mathbf{k} \sim 0$ (red) squares were obtained by Raman scattering (Ref. 14). The frequency scales in the insets have been expanded to show the effect of the *s*-*o* around *X*.



FIG. 5. (Color online) Phonon density of states of β -HgS, HgSe, and HgTe, (a), (b), and (c), respectively, as obtained with the ABINIT code used for Fig. 4. We have plotted the total DOSs and also their projection on the Hg and the anion atoms. Note that, in the case of β -HgS, the acoustic phonons correspond to nearly pure vibrations of Hg, the optic ones to nearly pure S vibrations. The (blue) vertical arrows mark the positions of the optical phonons at Γ .

as opposed to the negative one found for HgS. The situation is similar to that found for the copper halides: CuCl has a negative splitting whereas for CuBr and CuI it is positive.¹⁹ This fact has been attributed by Shindo *et al.* to the negative contribution of the copper 3*d* electron to the valence band splitting.²⁰ As already suggested above, a similar explanation holds for the Hg-chalcogenides: the negative splitting of the 5*d* electrons of Hg contributes, through hybridization, an amount to the valence splitting sufficient to reverse the 3*p* splitting of S but neither the 4*p* splitting of Se nor the 5*d* of tellurium. According to Table I, agreement between experimental and calculated values of Δ_0 is good. Notice that, because of the positive *s-o* splitting, HgSe and HgTe are zerogap semiconductors, like α tin.

In the process of performing these calculations, we also took a look at the terms linear in **k** which appear in the valence bands around Γ as a result of *s*-*o* interaction coupled with the lack of inversion symmetry. The existence of these terms was first reported by Dresselhaus,²⁷ and numerical calculations were performed by Cardona *et al.*,²⁸ by Surh *et al.*,⁴⁷ and more recently by Kim *et al.*⁴⁸ These terms arise in perturbation theory from *s*-*o* interaction between the valence bands and the *d* core electrons (Γ_{12} states) and back to the



FIG. 6. *Ab initio* calculations of the acoustic dispersion relations of HgSe performed with the ABINIT-LDA code, with (solid line) and without *s*-*o* (dashed line) interaction. The circles represent INS data taken from Ref. 56. We also performed calculations with the VASP-GGA code. The LA modes were found to be slightly lower in frequency (82 cm⁻¹ at *L*, 91 cm⁻¹ at *X*) as expected from the larger a_0 (see Table I). The TA modes barely changed.

valence band via $\mathbf{k} \cdot \mathbf{p}$.²⁸ In Ref. 28, *ab initio* calculations of these terms were presented for many III-V and II-VI zincblende-type semiconductors but not for the Hg-chalcogenides. These calculations may have been vitiated by "gap problems," but concerning the linear terms in \mathbf{k} , any such errors should be small because they do not involve the *p*-*s* gap. In Ref. 28, a semiempirical expression was given for the coefficient C_k of these linear terms. For the Hg-chalcogenides compounds, it can be written as [see Eq. (7.6) of Ref. 28]

$$C_k = -350 \frac{D_{d,c}}{E(\Gamma_8^{\nu}) - E_{d,c}(\Gamma_{12})} (\text{in meV Å}), \qquad (1)$$

where $\Delta_{d,s}$ is the *ds-o* splitting of the cation (for Hg 2.2 eV) and the energy denominator represents the gap between the Γ_8 valence band and the outermost *d*-states of the cation. In Eq. (1), we have neglected the contributions of the anion *d* electrons since they are absent for HgS and should be small for HgSe and HgTe. Equation (1) allows us to estimate C_k =-128 meV Å for the Hg-chalcogenides.

We now give some details concerning the definition of C_k . This parameter has been chosen such that the splitting for **k** along [100] is $\pm C_k k$ (the convention used for the sign is that of Ref. 28), each one of the two split states being a doublet for this **k** direction. For **k** along [111] two of the four Γ_8 states are not affected by linear **k** terms, whereas the other two split like $\pm \sqrt{2}C_k k$. For **k** along [110] the Γ_8 states split into 4, two like $\pm C_k k (1 + \sqrt{3}/2)^{1/2}$ and the other two like $\pm C_k k (1 - \sqrt{3}/2)^{1/2}$.

We display in Fig. 3 the valence bands of HgSe calculated in a small energy and k range so as to be able to observe the linear terms and thus determine C_k . The calculations were performed for **k** along [100] and [111] using the ABINIT code in the LDA approximation. Values of C_k were obtained through fits with a polynomial containing a linear and a qua-



FIG. 7. Phonon dispersion relations calculated with the ABINIT-LDA code for HgTe. The circles represent data from INS measurements (Ref. 51). Raman spectroscopy gives $TO(\Gamma)=116 \text{ cm}^{-1}$ and $LO(\Gamma)=139 \text{ cm}^{-1}$ (Ref. 57).

dratic (effective mass) term. Notice the splitting pattern mentioned above: for **k** along [100] the Γ_8 quadruplet splits into two doublets. For **k** along [111] a doublet does not split and exhibits no linear term (notice the small value of the linear fitting coefficient). The other doublet splits into $\pm \sqrt{2}C_kk$.



FIG. 8. (Color online) Sum [(black) solid line] and difference [(red) dashed line] of the two-phonon DOS of β -HgS, HgSe and HgTe. The wave vectors of the two phonons are taken to be equal (but opposite), as required by optical spectroscopies.

Experiment (cm ⁻¹)	Calculated (cm ⁻¹)	Assignment		
43	50, 65	2 TA		
110	110	LA+TA		
120	120	LA+TA		
125	130	LA+TA		
160 (weak shoulder)				
195	199, 200, 210, 220	O+TA		
260	260, 245, 275, 280	2 TA, LO+TA		
350	230 [exactly $2LO(\Gamma)$]	2 LO close to Γ , 2 LO(X)		
370	370	$2LO(L,\Sigma)$		

TABLE III. Structures observed in the second order Raman spectrum of HgSe (Ref. 58) compared with structures seen in the two-phonon DOS calculated on the basis of the band structure partly depicted (only the acoustic bands) in Fig. 7.

Similar patterns are found for HgS and HgTe. We shall not present here the corresponding figures. Instead, we shall list all of the values of C_k determined for our three materials with the ABINIT and the VASP codes.

We display in Table II the values of C_k for the three materials as found with the two computational codes from the [100] and the [111] splittings. The sign convention is the same as that of Refs. 28 and 47. These values are rather close to that estimated with Eq. (1) using for the *p*-*d* gap 6 and 2.2 eV for the *s*-*o* splittings of the 5*d* electrons of Hg(C_k = -128 meV Å). We have not found experimental values of C_k for these materials.

IV. LATTICE DYNAMICS

A few Inelastic Neutron Scattering (INS) determinations of the phonon dispersion relations of β -HgS,^{49,50} HgSe,⁵¹ and HgTe⁵¹ have been published. They have been compared with semiempirical calculations, which use fitted force constants. Here, we compare the measured points with *ab initio* calculations (ABINIT or VASP) and examine the effect of *s-o* interaction on these calculations. We start with β -HgS and display in Fig. 4 the phonon dispersion relations calculated with *s-o* interaction (solid line) and also without (dashed line). The effect of the *s-o* interaction (see inset for the TA phonons) is rather small, however it is possible to see that for the TA phonons, those mainly responsible for the C_v to be discussed later, the frequencies without *s-o* are about 5% lower than with *s-o*.

An interesting feature of Fig. 4 is the strong upwards bending of the optical bands away from $\mathbf{k}=0$, the opposite of

what happens for most tetrahedral semiconductors. This effect is due to the large mass difference of Hg and S: at the edge of the zone the "effective" mass which determines the frequency is basically that of sulfur, whereas at k=0 it is the somewhat smaller reduced mass. In order to obtain the calculated up-bending one must also invoke a strong force constant connecting the anions. The upward bending is reflected in the corresponding density of phonon states, shown in Fig. 5 together with those of HgSe and HgTe, to be discussed later. A direct consequence is that the TO Raman modes do not overlap the DOS continuum while the LO mode does. The width of the former should therefore not be affected by elastic scattering due to defects, such as isotopic fluctuations, whereas the LO modes, overlapping the background, should. This situation is opposite to that discussed in Ref. 52 for ZnSe, where isotopic disorder is shown to broaden the TO Raman phonons but not the LO phonons.

The up-bending of the dispersion relations are also important when considering confinement in quantum dots. These effects have been discussed for the case of CdS, a material for which the TO bulk bands bend up but the LO bands bend down.⁵³ No such experiments are available, to the best of our knowledge, for β -HgS, but quantum confinement in dots of this material should produce an increase in frequency as opposed to most other related materials. Up-bending is also found in the dispersion relations of HgSe and HgTe (see below), but not as strong as in the case of β -HgS, because the difference between anion and cation masses is not as large. A canonical example of up-bending phonon dispersion relations is found in rock-salt structure PbS. In this material, the LO phonons are found at 205 cm⁻¹ at the Γ point of the

TABLE IV. Isotopic fluctuation parameter $g(\times 10^5)$ for the elements under consideration, either as found in nature or prepared as equal mixtures of the two isotopes whose masses are given in brackets.

		Isoptope composition			
	Hg	S	Se	Te	
Natural	6.2	9.2	7.9	3.8	
50-50%	(198–204) 22	(32–34) 92	(76-80) 67	(124–130) 56	

	N _d Hg	N _{dCh}	$e_{0\mathrm{Hg}}$	$e_{0\mathrm{Ch}}$	$\Gamma_i \text{ (natural)} \ (\text{cm}^{-1})$	$\gamma_i(50-50\%) \ ({\rm cm}^{-1})$
β-HgS	0	0.015	0.15	0.99	0.02	0.2
HgSe	0.03	0.085	0.40	0.93	0.06	0.5
HgTe	0.02	0.05	0.53	0.85	0.02	0.2

TABLE V. Parameters needed to evaluate the disorder induced broadening of $LO(\Gamma)$ phonons with Eq. (2) and the isotope disorder induced broadenings obtained by this procedure.

BZ. A direct consequence of the up-bending is the fact that the corresponding peak in the second order Raman spectrum is not centered at $2 \times 105=210 \text{ cm}^{-1}$ but somewhat higher (228 cm⁻¹ at 12 K).⁵⁴ In contrast, for GaAs, a typical downbending case, the two LO-phonon peak is shifted down by as much as 5 cm⁻¹ with respect to twice the frequency of the LO phonons at Γ .⁵⁵ Effects of the upwards bending on the phonon line widths will be discussed below.

Figures 6 and 7 display the phonon dispersion relations of HgSe and HgTe, respectively, as calculated with the ABINIT-LDA code. In Fig. 6, we only show the acoustic phonon branches and the corresponding INS data, because *ab initio* calculations, using the VASP-ASA code, have already been reported by Łazewski *et al.*⁵⁶ The main purpose of this figure is to visualize the effect of the *s-o* interaction which slightly raises the frequencies of the TA modes and, as we shall see below, lowers the calculated maximum of C_v/T^3 .

The optical branches calculated in Ref. 56 for HgSe also bend upwards when moving away from Γ , but not as much as in Fig. 4. This is expected because for HgSe anion and cation masses differ less than in the case of β -HgS. Correspondingly, and as shown in Fig. 5, the optical phonon bands of HgSe are narrower than those of β -HgS. Moreover, the projected densities of states of these phonons correspond about 90% to Se and 10% to Hg.

Figure 7 displays the full acoustic and optic dispersion relations for HgTe together with INS data. In this case, the upward bending of the optical bands is rather small, as corresponds to cation and anion masses which differ by less than a factor of two. Because of the weak upward bending, the optical DOS band is even narrower than for HgSe (see Fig. 5).

For the sake of completeness, we include a figure with the two-phonon DOS, which applies to second order optical processes in which either two phonons are created (sum processes) or one is created and the other destroyed (difference processes). After multiplication by the appropriate matrix elements (which we have not calculated in this work), the two-phonon DOS should describe the corresponding IR and Raman spectra. Even without multiplication by the matrix elements, structure in the two-phonon DOS should help to interpret the corresponding spectra.

There is only scarce information about the second order Raman and IR spectra of the materials under consideration. We have already mentioned the 2LO structure seen for β -HgS, which is slightly upshifted with respect to twice the frequency of the LO(Γ) phonons. Szuszkiewicz *et al.* have reported Raman spectra for HgSe with structures which can be attributed to two-phonon sums.⁵⁸ A rather detailed assignment is now possible using the DOS of Fig. 8. The resolution of their measurements is better than 2 $\,\mathrm{cm}^{-1}$. The full width at half maximum (FWHM) of the observed TO phonon is ~ 5 cm⁻¹. After correcting for instrumental resolution,⁵⁹ the FWHM becomes $\sim 4.2 \text{ cm}^{-1}$. We shall discuss below a possible isotope disorder contribution to this width.⁶⁰ The temperature at which the spectra were measured (90 K) most likely precludes the observation of difference processes. The measurements were performed in back-scattering using a [110] surface of cleavage. Under these conditions, the selection rules prevent the observation of single LO phonons. The single TO mode is seen at 133 cm⁻¹. We show in Table III the frequencies of the observed "two-phonon" peaks and their assignment to structures in Fig. 8. We also have made an attempt, as given in Table III, to assign these structures to the dispersion relations calculated for HgSe. The exact overtones of the phonons at Γ [2TO(Γ)=264 cm⁻¹ and 2LO(Γ) =330 cm⁻¹] are not clearly observed: 2TO(Γ) corresponds to a threshold and not a peak (see Fig. 8), $2LO(\Gamma)$ should be slightly shifted upwards because of the up-bending and may overlap with the broad band that we have assigned to 2 LO(X), as indicated in Table III.

Linewidths of Γ phonons and isotope disorder contribution to their broadening

We have found reliable data for these linewidths in the cases of HgSe (Ref. 58) and HgTe.⁵⁷ The resolution reported in Ref. 57 for HgTe is 1 cm⁻¹ (FWHM). Correspondingly, no resolution correction is needed for the measured linewidths which are 5 cm⁻¹ for the TO(Γ) phonon and 4.5 cm⁻¹ for LO(Γ), at 1.6 K.⁵⁹ In the work of Szuszkiewicz on HgSe, the resolution is quoted to have been better than 2 cm^{-1} , standard deconvolution⁵⁹ of the observed line width of the TO phonons (5 $\,\mathrm{cm}^{-1}$) brings the measured FWHM to about 4.2 cm⁻¹. These materials being semimetallic, there should be some contribution of transitions from the occupied to the empty valence bands to the phonon linewidths. An investigation of this contribution has not been performed. The temperature dependence due to anharmonic decay into two phonons,⁶¹ has not been investigated for these materials, neither experimentally nor theoretically. Nevertheless, extant data, even those obtained at the lowest temperatures, are affected by anharmonic effects related to the zero-point vibrations.⁶⁰ These zero point effects, however, are typically less than 1 cm⁻¹ (FWHM) for most zincblende-type semiconductors.⁶¹ We thus surmise that the electron-phonon coupling across the Γ_8 valence bands by either the LO or the TO Γ phonons must give a sizeable contribution to their



FIG. 9. (Color online) Enthalpy of the three phases of HgS calculated with the ABINIT-LDA code with *s*-*o* interaction. Cinnabar and rock-salt phases are basically degenerate (within any reasonable error estimates) at p=0, although formally, their enthalpies cross at ~ 0.7 GPa. The cinnabar \rightarrow rock salt transition is predicted to occur at ~ 22 GPa. Experimental data place it at 20.5 GPa (Ref. 64).

FWHM. Before we assign to electron-phonon coupling, the difference between the observed FWHM ($\sim 5 \text{ cm}^{-1}$) and the expected anharmonic effect ($\sim 1 \text{ cm}^{-1}$), we must, however, estimate the contribution of the isotopic disorder in materials grown from the natural elements. This can be done with the expression⁶⁰

$$\Gamma_{i} = \frac{\pi \omega_{0}^{2}}{6} [g_{\rm Hg} N_{d\rm Hg}(\omega_{0}) |e_{\rm 0Hg}|^{2} + g_{\rm Ch} N_{d\rm Ch}(\omega_{0}) |e_{\rm 0Ch}|^{2}], \quad (2)$$

where Γ_i represents the contribution of isotopic disorder to the FWHM of either the TO or the LO phonon, g_{Hg} and g_{Ch} the isotopic mass fluctuation parameter of either Hg or the chalcogen, respectively,

$$g_{\rm Hg,Ch} = \frac{\langle M_{\rm Hg,Ch}^2 \rangle - \langle M_{\rm Hg,Ch} \rangle^2}{\langle M_{\rm Hg,Ch} \rangle^2}.$$
 (3)

In Eq. (2), N_{dHg} and N_{dCh} are the densities of one-phonon states projected on Hg and the chalcogen, respectively, while e_{0Hg} and e_{0Ch} are the corresponding eigenvector amplitudes for the phonons at Γ . We list in Table IV the values of g for the natural elements under consideration and for artificial isotopic mixtures conceived so as to maximize g taking into account the price of the isotopes. The parameters needed to evaluate Eq. (2) are listed in Table V.

We thus conclude that the observed FWHM of $TO(\Gamma)$ and $LO(\Gamma)$ phonons in HgSe and HgTe, between 3 and 4 cm⁻¹ after subtracting anharmonic effects, must be due to electronphonon interaction between full and empty valence bands. The effect can be compared to that observed in heavily doped Si and Ge. For Si with 6×10^{19} cm⁻³ holes a contribution of about 5 cm⁻¹ to the FWHM is found.⁶²

V. PRESSURE EFFECTS ON THE STRUCTURE AND LATTICE DYNAMICS

This investigation was triggered by our interest in calculating the specific heats of the Hg-chalcogenides crystallized in the rock-salt structure (see next section). These materials undergo phase transitions at relatively low pressures.⁶³ While



FIG. 10. (Color online) Enthalpy of HgSe calculated with the ABINIT-LDA code including *s*-*o* splitting. The arrows signal the calculated zincblende \rightarrow cinnabara and cinnabar \rightarrow rocksalt transitions, which are found experimentally to take place at 0.4 and 14.6 GPa (Refs. 65).

there is considerable experimental data on these transitions (see Ref. 63), no *ab initio* calculations seem to have been reported. We have performed enthalpy minimizations vs pressure for HgS and HgTe considering the three most conspicuous structures of these materials: cinnabar, zincblende, and rock salt. Orthorhombic and bcc structures have also been observed experimentally for HgTe, but we have not included them in our calculations.

We show in Fig. 9 the pressure dependence of the enthalpy of the three phases mentioned, as calculated for HgS. The cinnabar phase (α -HgS) is found, within estimated error, to have the same enthalpy as the zincblende phase (β -HgS), a fact that may explain the ease to obtain both phases in the laboratory, although the growth of the zincblende phase requires the addition of a small amount of Fe.¹⁴

Notice, however, that the zincblende phase is calculated to be the stable one at zero pressure, but with an enthalpy only about ~0.0003 Ha below the cinnabar phase. We do not believe that this minute enthalpy difference is significant. The cinnabar phase is calculated to be stable between 0.7 and 22.2 GPa, at which pressure the NaCl phase should appear. This is in reasonable agreement with reports of a phase transition at 20.5 GPa.⁶⁴ In the next section, we shall present calculations of the specific heat of the rock-salt phase of HgS in the stable pressure region, so as to ascertain the effect of *s-o* interaction on this phase.

In Figures 10 and 11, we present similar calculations for the enthalpies of HgSe and HgTe vs pressure.

The calculations were also performed without *s*-*o* interaction and we found that the presence of this interaction slightly improves agreement with experimental results which are 1.5 GPa for the zincblende \rightarrow cinnabar transition (as compared with 1.3 GPa in Fig. 11, 1.0 without *s*-*o* interaction. 8 GPa for the cinnabar \rightarrow rock salt transition has been also observed.⁶⁶ For reasons unbeknownst to us, it is not reproduced in our calculations.

As a complement to this pressure investigation, we calculated the pressure dependence of the mode frequencies of the LO(Γ) and TO(Γ) phonons for zincblende HgTe. The results, as obtained with the ABINIT-LDA code with and without *s-o* interaction, are shown in Fig. 12

They can be expressed in terms of the usual Grüneisen parameters using as a bulk modulus B=42.2 GPa (Ref. 25)



FIG. 11. (Color online) Enthalpy of HgTe calculated with the ABINIT-LDA code including *s-o* splitting. The arrow signals the calculated zincblende \rightarrow cinnabar transition which is found experimentally to take place at 1.5 GPa (Ref. 66).

$$\gamma_{\rm LO} = -\frac{d \ln \omega_{\rm LO}}{d \ln V} = 1.0; \quad \gamma_{\rm TO} = -\frac{d \ln \omega_{\rm TO}}{d \ln V} = 1.22, \quad (4)$$

using Fig. 12 it is also possible to determine a Grüneisen parameter for the transverse effective charge e^* as defined with the equation

$$\omega_{\rm LO}^2 - \omega_{\rm TO}^2 = \frac{4\pi N e^{*2}}{M\epsilon_{\infty}}.$$
 (5)

The calculation of a Grüneisen parameter [i.e., a logarithmic derivative with respect to the volume), like in Eq. (4)] for e^* from the data of Fig. 12 requires, beside the bulk modulus, the knowledge of the Grüneisen parameter $\gamma_{\epsilon_{\infty}}$ of the electronic dielectric constant ϵ_{∞} .⁶⁷ For lack of better information, we take a value of $\gamma_{\epsilon_{\infty}}$ =-1, typical of that found for most tetrahedral semiconductors⁶⁸ and find for the Grüneisen parameter of e^* : γ^* =-0.65. This negative sign implies that e^* increases with increasing volume. It is usually interpreted to signal an increase in ionicity with increasing volume, a result that is common to most zincblende-type semiconductors with the exception of SiC (see Table IV in Ref. 67).



FIG. 12. (Color online) Pressure dependence of the frequencies of the LO(Γ) (upper curves) and TO(Γ) phonons (lower curves) of HgTe calculated with the ABINIT code with and without *s-o* interaction as indicated.

VI. SPECIFIC HEAT

The specific heat C_p of the crystalline samples was measured for the three chalcogenides between 2 and 280 K with a PPMS system (Quantum Design, 6325 Lusk Boulevard, San Diego CA (Quantum Design, 6325 Lusk Boulevard, San Diego, CA), as described before.^{3–5} The powder samples of β -HgS were sealed into glass flasks under \approx 900 mbar ⁴He gas to ensure thermal contact at low temperatures. The heat capacities of such samples were measured using a quasiadiabatic heat pulse method with a home-built Nernst calorimeter similar to that described Schnelle and Gmelin.⁶⁹ The heat capacities of the glass sample container, the sapphire platform and a minute amount of Apiezon N vacuum grease used to thermally couple the sample to the platform were measured separately and subtracted.

The specific heat of HgSe above 80 K has been measured by Gul'tyaev and Petrov.⁷⁰ Low temperature specific heats of HgTe have been determined by Collins *et al.* and subsequently for β -HgS, HgSe, and HgTe by Khattak and co-workers.^{71,72} Our data for HgSe and HgTe agree very well with those measured previously. We find, however, very large differences at temperatures around 7.5 K where the maximum in C_p/T^3 occurs for our β -HgS data compared to those of Khattak *et al.*, our data being a factor ~2 larger.

Calculations of C_v were performed by integrating the calculated phonon DOS with the standard expression (we have already mentioned that in the region of interest $C_p \approx C_v$). The calculations were performed both, with the ABINIT-LDA and the VASP-GGA codes, with and without *s-o* interaction. The effect of *s-o* interaction was basically the same for the two codes: the *s-o* interaction lowered the maximum of C_v/T^3 by about 10%. We shall present here the results obtained with only one of the codes (except for β -HgS) but both, with and without the *s-o* interaction.

We show in Fig. 13(a) the experimental values of C_v/T^3 for two sample of β -HgS, a natural crystal and a high-purity powder. The natural crystal was found by chemical microanalysis to have an Fe content of 1.1 mass-%. For comparison, we also display measurements by Khattak *et al.*⁷² The upturn at low temperatures observed for the natural crystal can be ascribed to a linear term in the specific heat (Sommerfeld term) due to metallic behavior. To correct for this term, we subtracted a linear contribution γ =20 mJ/mol K2) from the experimental data. Corrections for an FeS content using the data by Kobayashi *et al.*⁷³ lead to a negligible change in the data and is not shown here.

Fig. 13(b) displays calculated VASP and ABINIT values of C_v/T^3 for β -HgS obtained with and without *s*-*o* interaction. The agreement between theoretical and experimental data is reasonably good.

Figure 14 shows C_p/T^3 measured by us and the values calculated from data by Gul'tyaev and Petrov,⁷⁰ compared with ABINIT-LDA calculations which include *s-o* coupling and also with VASP-GGA calculations with and without *s-o* interaction. The latter corroborate the general trend that *s-o* interaction decreases C_v by about 10%, a fact that corresponds to an increase in the TA phonon frequencies. The VASP calculations agree better with the experiment than the ABINIT ones but the excellent agreement with experimental data found for



FIG. 13. (Color online) (a) C_v/T^3 measured on a mineral and a powder sample of β -HgS. Data by Khattak *et al.* are also given for comparison (Ref. 72). The solid line represents the mineral data minus a linear contribution as indicated in the inset. (b) Same mineral data as before compared with the results of four calculations as indicated in the inset.

the VASP calculations without *s-o* interaction should, of course, be considered fortuitous. The poor agreement of the ABINIT calculations with the experimental results corresponds to the poor agreement found in Fig. 6 for the TA phonons along the Γ -X direction. The reason for the poor agreement found in this case is not known to us in detail. We presume it is due to inaccuracies in the *ab initio* pseudopotentials employed.

At high temperatures, contributions linear in temperature due to thermal expansion have been added to the ABINIT data (see insets in Figs. 14 and 15). While the linear term for HgTe(9.08×10^{-3} J/mol K²T) is in good agreement with lit-



FIG. 14. (Color online) C_v/T^3 calculated for HgSe vs *T* with the VASP code with (red dashed line) and without (blue solid line) (*s*-*o* interaction) and (insert) with the ABINIT code (only with *s*-*o* interaction). The effect of the latter is similar to that found for the VASP calculations). Also, C_p/T^3 measured by us and measurements by Gul'tyaev and Petrov and Khattak *et al.* are presented (Refs. 70 and 72). The inset depicts $C_v \approx C_p$ obtained by ABINIT with *s*-*o* interaction included. A contribution linear in temperature, 4×10^{-3} J/mol K²T, due to thermal expansion, has been added to the calculated data.



FIG. 15. (Color online) C_v/T^3 vs *T* for HgTe calculated with both, ABINIT-LDA and VASP-GGA codes including and not including *s-o* interaction compared with our experimental data. The inset depicts $C_v \approx C_p$ obtained by ABINIT with *s-o* interaction included. A contribution linear in temperature, 9.08×10^{-3} J/mol K²*T* (Ref. 25), due to thermal expansion has been added to the calculated data.

erature data, that for HgSe($\sim 4 \times 10^{-3}$ J/mol K²*T*) is smaller by a factor of 3.5 than data reported in the literature.²⁵ Our experimental results are, however, in very good agreement with data reported by Gul'tyaev and Petrov.⁷⁰

We would like to recall the fact that the s-o effects calculated here for materials with zincblende structure are opposite to those calculated in Romero et al.⁷ for the lead chalcogenides, which crystallize in the rock-salt structure. In order to investigate whether this difference is characteristic of the crystal structures, we calculated the dispersion relations of rock-salt type HgS, which, as shown in Fig. 9, is stable above 21.6 GPa (the experimental phase transition pressure is 20.5 GPa).⁶⁴ There are, of course, no measurements for HgS in the rock-salt phase but we show in Fig. 16 the C_n/T^3 vs T calculated with the VASP-GGA code. In order to stabilize the rock-salt structure, we performed the calculations assuming a pressure of 25 GPa applied to the material. The s-o interaction induces a minute decrease of C_n/T^3 (less than 1%) which can be assumed to be within the possible error of the calculations using different potentials. The change in the sign of the effect when going from zincblende to rock salt, found in the calculations going from the Hg-



FIG. 16. (Color online) Calculations, using the VASP-GGA code with (red) dashed line and without (solid line) *s*-*o* interaction, of HgS at a pressure of ~25 GPa (rock-salt structure). Notice that the *s*-*o* interaction lowers C_v/T^3 by an amount that is insignificant with respect to uncertainties in the computational method. The inset has been added in order to clearly illustrate the small effect of *s*-*o* interaction.

chalcogenides to the Pb chalcogenides, is not confirmed by the calculations performed for HgS for the rock-salt structure.

The difference between zincblende and rock salt, however, has the right sign. We thus conclude that the reason for the change in sign of the effect of s-o interaction, from the Hg to the Pb chalcogenides is likely to be related to the change in structure.

The reader may wonder why the maximum of C_v/T^3 in Fig. 16 is smaller by a factor of seven than calculated for the zb phase. In order to ascertain the reason for this difference we calculated the phonon DOS for the rs phase. We found the TA peak to occur at 62 cm⁻¹, a factor of 2.3 larger than that shown in Fig. 13(b) for the zb phase. This difference can account for the lower maximum in C_v/T^3 calculated for the rs phase.

VII. CONCLUSIONS

The original motivation of the present work was to perform accurate measurements of the specific heat of β -HgS, HgSe, and HgTe at low temperatures and to compare them with ab initio calculations of the band structures of these materials, thought to be semimetals with the β -tin inverted valence band structure. A prerequisite was the calculation of the electronic band structures of these materials. While doing so, we found a number of thus far unknown details and their interpretation. We mention first that there is a significant hybridization of the *d* core electrons of Hg with the (chalcogen) *p*-like valence bands which results in a negative contribution to their s-o splitting at Γ . Consequently, for β -HgS this s-o splitting even reverses its sign. Thus, the material, believed to be a semimetal, becomes a semiconductor with a gap similar to the magnitude of the Γs -o splitting (~0.1 eV). The s-o splittings calculated for the three materials compare satisfactorily with experimental results, whereas the negative E_0 gaps are larger in magnitude as a consequence of the socalled gap problem associated with LDA. The presence of s-o interaction in our calculations allows us to determine the coefficients of the linear terms of the Γ_8 valence bands. They compare well with estimates based on a semiempirical interpolation formula used in the past for other zincblende-type semiconductors.

We next proceeded to the calculation of the phonon dispersion relations and comparison with experimental results based on Raman and INS spectroscopies. We performed these calculations using existing computer codes and investigated the effect of s-o interaction which turned out, in the case of the acoustic phonons, to have a sign opposite to that found for the rock-salt structure of the Pb chalcogenides. This effect resulted in changes by about 10% in the values of the calculated maximum of C_{ν}/T^3 vs T: the s-o splitting lowers the value at this maximum, an effect that is opposite to that found for the rock-salt structure lead chalcogenides. An argument was found to correlate the sign of this effect with the structure of the materials by calculating C_v/T^3 for HgS in the rock-salt structure. From the calculated phonon dispersion relations, we obtained the phonon densities of states and their atomic projections. They were used for an analysis of measured phonon line widths and the corresponding effects of isotopic composition. We concluded that there must be a large contribution of the electron-phonon interaction across the Γ_8 bands to the line widths of the Raman phonons of HgSe and HgTe. We also calculated two-phonon densities of states in order to interpret extant and future second order Raman spectra. Finally, we investigated a few effects of hydrostatic pressure on the Raman phonons of these materials, in particular HgTe. We calculated the Grüneisen parameters of LO and TO phonons and also of the transverse (Born) effective charge. The latter was found to be γ^* =-0.65 for HgTe, a value similar to that found for most other zincblende-type semiconductors. the sign of γ^* has been interpreted as signaling a decrease in the ionicity with increasing pressure.

Note added in proof. The GW correction of the gap problem reverses the sign of E_0 given in Table I making $E_0 = +0.40$ eV. We are grateful to A. Svane for communicating unpublished results.

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- ${}^{1}C_{p}$ represents the experimentally measured value of the heat capacity at constant pressure. The *ab initio* calculations yield C_{v} , i.e., the heat capacity at constant volume. In the temperature region considered here, $C_{v} \approx C_{p}$.
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